

Silyl alkyl esters of anthracene- and phenanthrene
carboxylic acids

The present invention relates to silyl alkyl esters, particularly of anthracene- and phenanthrene carboxylic acids, a method for their preparation, compositions and polysiloxane compositions, which contain the silyl alkyl esters according to the invention and can be used particularly in the semiconductor industry for the production of anti-reflective layers for photolithographic applications.

Progress in the semiconductor industry is based on the continuing development of new generations of integrated circuits which need to exhibit greater efficiency and functionality with ever-decreasing dimensions. This leads to new challenges in many processes conventionally used in semiconductor production. One of the most important of these processes is photolithography.

It has long been recognized that linewidth variations in

patterns produced by photolithography can result from optical interference from light reflected by an underlying layer of a semiconductor wafer. Variations in thickness of the photoresist layer due to the topography of the subjacent substrate or layer also induce linewidth variations. In the prior art, anti-reflective layers applied under a photoresist layer have been used to prevent interference from reflection of the transmitted light irradiating beam. In addition, to some extent such layers are capable of planarizing the wafer topography and thus of producing a more uniform thickness of the photoresist layer, which contributes to less linewidth variations.

Organic polymers have already been employed in anti-reflective layers during irradiation of photoresist materials at the conventional wavelengths (365 nm, 436 nm) and also the shorter wavelengths (248 nm) used recently. However the fact that these share many chemical properties with the organic photoresists can limit usable process sequences. Furthermore they may intermix with photoresist layers, and additives have conventionally been introduced to avoid this. An example of suitable additives are thermosetting binders, as described in US-A-5,693,691.

Silicon oxynitride is another material that has been used in anti-reflective layers and it works mainly by a destructive interference rather than by absorption, thus making it necessary to control very strictly the coating thickness. This makes it difficult to use it if highly uneven topographies are intended to be coated. Since silicon oxynitride is typically deposited by chemical vapour deposition, while photoresist layers are typically applied using spin-techniques, the complexity of the semiconductor production is even increased.

Another class of materials are so-called spin-on-glass compositions. For example, US-A-5,100,503 discloses a solution of cross-linked polyorganosiloxane containing an

inorganic pigment, e.g. TiO_2 amongst others, and an adhesion promoter. Although planarizing layers can be produced with such polyorganosiloxane-pigment combinations, they are not optimally suitable for exposure in the short-wave ultraviolet range, e.g. at 248 and 193 nm, which is increasingly used for the production of semiconductors with very small dimensions. Furthermore, not all inorganic pigments are suitable for the incorporation into different spin-on-glass compositions.

WO-A-00/77575 discloses spin-on-glass compositions containing a siloxane polymer and an incorporatable absorbing organic compound. For example, these compounds have a chromophore group with 1 to 3 condensed or non-condensed benzene rings, and a reactive group bonded to the chromophore group, e.g. hydroxy groups, amine groups, carboxylic acid groups and substituted silyl groups, with silicon bonded to one, two, or three alkoxy groups or to halogen. The reactive group may be directly bonded to the chromophore or the reactive group may be attached to the chromophore through a hydrocarbon bridge. These compounds, e.g. 9-anthracene carboxylic acid triethoxysilyl methyl ester, show strong absorption at wavelengths in the ultraviolet range, e.g. 193 nm, 248 nm or 365 nm, and are therefore also suitable for modern photolithographic processes. However, further improvement is required with regard to their hydrolytic and thermal stability, which is essential for the yield of technical syntheses, subsequent storage and the production of coating materials suitable for the spin-on-process, which normally takes place in aqueous-alcoholic media.

WO-A-00/77575 also discloses a method for the production of 9-anthracene carboxylic acid triethoxysilyl methyl ester, in which 9-anthracene carboxylic acid, chloromethyl-triethoxysilane, triethylamine and a solvent are mixed and heated under reflux. On cooling of the reaction mixture a solid forms and a supernatant solution, with the latter containing the desired end product. However, this method requires further improvement with regard to yield and purity

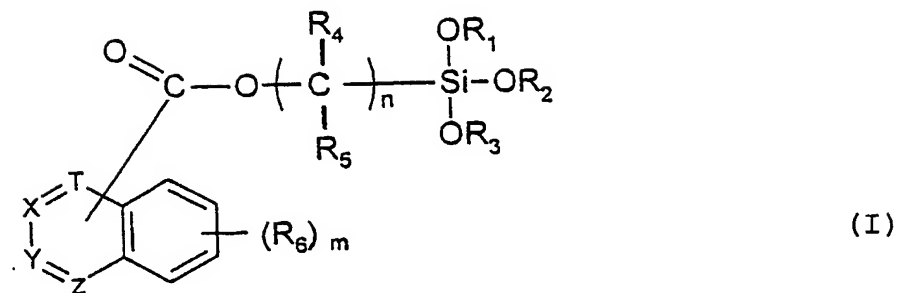
of the desired end product. For example, the synthesis of 9-anthracene carboxylic acid triethoxysilyl methyl ester yields a dark-brown oily liquid which contains only approx. 25 wt.-% silyl alkyl ester. In addition, the working-up of the
5 reaction mixture is time-consuming as, in addition to decanting and removal of the solvent, a column chromatographic purification is necessary. Because of the stringent purity requirements in the semiconductor industry and for economic considerations, it is generally desirable to
10 be able to isolate the above compounds in a higher purity and yield.

The object of the invention is therefore, firstly, to provide a light-absorbing compound, which has a strong absorption at
15 wavelengths in the ultraviolet range, e.g. 193 and 248 nm, and is more stable with regard to hydrolytic and/or thermal influences than known compounds containing mono- or polynuclear aromatic hydrocarbon radicals as chromophore groups.

20 A further object of the invention is to provide a method for preparing silyl alkyl esters containing polynuclear condensed aromatic hydrocarbons as a chromophore group, with which an even higher yield of the desired final product can be
25 obtained than with conventional methods. In particular the final product should have even higher purity than is possible with known methods, in spite of a simpler working-up of the reaction mixture.

30 These objects are achieved by the silyl alkyl esters according to Claims 1 to 10 and by the method according to Claims 37 to 45.

The silyl alkyl ester according to the invention having the
35 formula (I)



is characterized in that

R_1 , R_2 , and R_3 are equal or different and represent alkyl, aryl and heteroaryl,

R_4 and R_5 are equal or different and represent hydrogen,

5 halogen, alkyl, aryl and heteroaryl,

n is an integer from 1 to 10,

R_6 is a substituent selected from halogen, alkyl, aryl, heteroaryl, hydroxy, alkoxy, aryloxy, substituted and unsubstituted amino group, carboxy group, carboxylic acid ester group, carboxylic acid amide group, sulfonic acid group, sulfonic acid ester group, sulfonyl, thio, thio ether and nitro,

m is an integer from 0 to 4,

T , X , Y and Z each represent carbon,

15 a benzo group, which is being substituted m -fold with R_6 or is unsubstituted, is condensed on one of the bonds $T-X$, $X-Y$ or $Y-Z$ to form a trinuclear aromatic ring system, wherein the silyl alkyl ester group is substituted at the middle ring of said trinuclear aromatic ring system,

20 with a silyl alkyl ester of the formula (I) being excluded, in which

R_1 , R_2 , and R_3 each represent ethyl,

R_4 and R_5 each represent hydrogen,

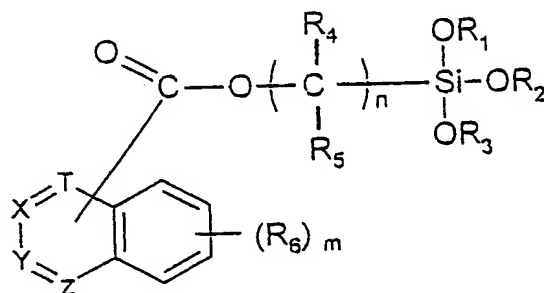
$n = 1$,

25 $m = 0$,

an unsubstituted benzo group is condensed on the $X-Y$ bond

and the silyl ester group is substituted at the 9-carbon atom of the trinuclear aromatic ring system.

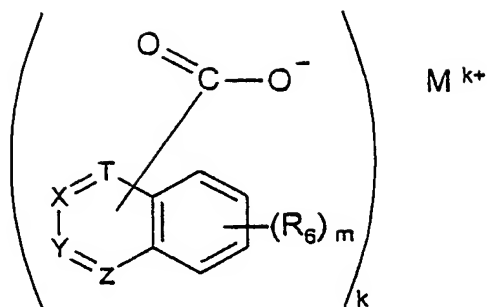
The method according to the invention for the preparation of a silyl alkyl ester having the formula (I)



(I)

in which

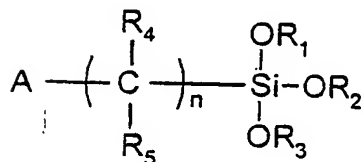
- 5 R_1 , R_2 , and R_3 are equal or different and represent alkyl, aryl and heteroaryl,
 R_4 and R_5 are equal or different and represent hydrogen, halogen, alkyl, aryl and heteroaryl,
 n is an integer from 1 to 10,
 10 R_6 is a substituent selected from halogen, alkyl, aryl, heteroaryl, hydroxy, alkoxy, arylother, substituted and unsubstituted amino group, carboxy group, carboxylic acid ester group, carboxylic acid amide group, sulfonic acid group, sulfonic acid ester group, sulfonyl, thio, thioether
 15 and nitro,
 m is an integer from 0 to 4,
 T , X , Y and Z each represent carbon,
 a benzo group, which is substituted m -fold with R_6 or is unsubstituted, is condensed on one of the bonds $T-X$, $X-Y$ or
 20 $Y-Z$ to form a trinuclear aromatic ring system, wherein the silyl alkyl ester group is substituted at the middle ring of said trinuclear aromatic ring system,
 is characterized in that a carboxylic acid salt having the formula (II)



(II)

in which k is an integer from 1 to 4 and M is a metal,
is reacted with an organosilicon compound of the formula
(III)

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(III)

in which A is a nucleofugal leaving group.

A further object of the invention is a composition,
particularly a solution, which contains at least one silyl
10 alkyl ester according to the invention and at least one

additional reactive silane.

Another object of the invention is a method for the preparation of a polysiloxane composition, particularly a polysiloxane solution, wherein the composition according to the invention is provided and the silyl alkyl ester is condensed with the reactive silane.

Yet another object is a polysiloxane composition, particularly a polysiloxane solution which is obtainable by the above method.

The invention also relates to a coated substrate, which is obtainable by a method comprising applying the composition or polysiloxane composition according to the invention to the substrate (e.g. a semiconductor device or silicon wafer) and heating the substrate with the composition or polysiloxane composition applied thereon.

Furthermore, the invention relates to the use of the above composition and the polysiloxane composition for preparation of a coating on any substrate, particularly for the preparation of anti-reflective layers on a semiconductor device or a silicon wafer.

Finally, the invention relates to an improved method for the photolithographic production of an integrated circuit pattern.

Surprisingly, it has been found that the silyl alkyl esters according to the invention are more stable with regard to hydrolytic and thermal influences than conventionally synthesized compounds, which contain polynuclear condensed aromatic hydrocarbon radicals as chromophore groups. In addition they have a strong light-absorption, particularly in the ultraviolet range, e.g. at wavelengths of 193 and 248 nm. This enables their use in anti-reflective coating materials for photolithographic applications, particularly

in the production of semiconductor devices which are exposed to the above wavelengths.

The compounds according to the invention represented by
5 Formula (I) are silyl alkyl esters of carboxylic acids of a trinuclear condensed aromatic hydrocarbon.

The radicals R_1 , R_2 , and R_3 are equal or different and represent alkyl, aryl and heteroaryl. In a preferred
10 embodiment the radicals R_1 , R_2 , and R_3 each represent alkyl, yet more preferably linear or branched alkyl having 1 to 6 carbon atoms. In particular, the radicals R_1 , R_2 , and R_3 are selected, independently of one another, from methyl, ethyl, n-propyl, iso-propyl, butyl, 2-methylpropyl (iso-butyl), 1-
15 methylpropyl (sec.-butyl), and 1,1-dimethylethyl (tert.-butyl). It is preferred that the radicals R_1 , R_2 , and R_3 each represent methyl or ethyl, with ethyl being most preferred.

The radicals R_4 and R_5 in the bridging group are equal or
20 different and represent hydrogen, halogen, alkyl, aryl and heteroaryl. Preferably R_4 and R_5 each represent hydrogen.

n is an integer from 1 to 10, preferably 1 to 5, more preferably 1 to 3, yet more preferably 1 or 3 and most
25 preferably 3.

R_6 is a substituent selected from halogen, alkyl, aryl, heteroaryl, hydroxy, alkoxy, arylother, substituted and unsubstituted amino group, carboxy group, carboxylic acid
30 ester group, carboxylic acid amide group, sulfonic acid group, sulfonic acid ester group, sulfonyl, thio, thio ether and nitro. Preferred substituents are halogen, alkyl, aryl, hydroxy, alkoxy and arylother, with halogen being particularly preferred.

35 Each of said hydrocarbon-containing radicals R_1 to R_6 can be branched or unbranched and carry further substituents, e.g. alkyl, which in turn can be branched or unbranched.

m is an integer from 0 to 4, preferably from 0 to 2 and particularly from 0 to 1, with $m = 0$ being the most preferred.

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T, X, Y, and Z each represent carbon. On one of the bonds T-X, X-Y, or Y-Z formed by these carbon atoms, a further benzo group, which is substituted m-fold with R_6 or is unsubstituted, is condensed to form a trinuclear aromatic ring system. The condensation with the unsubstituted or m-fold with R_6 substituted benzo group can be either linear on the X-Y bond, or angular on either the T-X or Y-Z bond. Accordingly, the trinuclear condensed aromatic hydrocarbon radical represents an anthracene or phenanthrene moiety.

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The groups already mentioned above can be used as substituent R_6 of the benzo group condensed on the T-X, X-Y or Y-Z bonds which groups, however, can be selected independently of one another for the respective rings of the trinuclear condensed aromatic hydrocarbon radical.

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The silyl alkyl ester group is substituted at the middle ring, i.e. bonded either on the 9- or on the 10-carbon atom of the anthracene or phenanthrene moiety. Preferably the silyl alkyl ester group is bonded on the 9-carbon atom.

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The other respective position of the middle ring can likewise carry a substituent which can preferably be selected from the radicals already listed for the substituent R_6 . It is preferred that the anthracene or phenanthrene moiety is substituted on the 10-carbon atom.

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In a yet more preferred embodiment of the silyl alkyl ester according to the invention, an unsubstituted benzo group is condensed either on the X-Y bond or on either the T-X or Y-Z bond, and in each case $m = 0$. This means that the whole condensed aromatic ring system, with the exception of the silyl alkyl ester group on the 9- or 10-carbon atom, carries

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no substituents.

Therefore, in the formula (I) an anthracene or phenanthrene radical, which is substituted on the respective 9-carbon atom with the silyl alkyl ester group is most preferred as trinuclear condensed aromatic hydrocarbon radical. In the case of an anthracene radical the compounds 9-anthracene carboxylic acid-3-(trimethoxysilyl) propyl ester or 9-anthracene carboxylic acid-3-(triethoxysilyl) propyl ester are examples of particularly preferred silyl alkyl esters according to the invention. In the case of a phenanthrene radical 9-phenanthrene carboxylic acid (trimethoxysilyl) methyl ester, 9-phenanthrene carboxylic acid (triethoxysilyl)- methyl ester, 9-phenanthrene carboxylic acid (triethoxysilyl)- propyl ester, or 9-phenanthrene carboxylic acid (triethoxysilyl)- propyl ester are examples of particularly preferred silyl alkyl esters according to the invention.

From the silyl alkyl esters according to the invention, a compound with the formula (I) is excluded, wherein an unsubstituted benzo group is condensed on the X-Y bond, $n = 1$, $m = 0$, R_1 , R_2 , and R_3 each represent ethyl, R_4 and R_5 each represent hydrogen and the silyl ester group is substituted on the 9-carbon atom of the trinuclear aromatic ring system. This is 9-anthracene carboxylic acid(triethoxysilyl) methyl ester.

The silyl alkyl esters according to the invention can generally be used as light-absorbing agents, particularly for absorption in the ultraviolet light range. Surprisingly, silyl alkyl esters according to the invention, in which the trinuclear condensed aromatic hydrocarbon radical is a phenanthrene radical, show particularly wide absorption in the ultraviolet light range. They absorb not only at a wavelength of 248 nm like the compounds based on an anthracene radical, but also at 193 nm. Generally this enables them to be used as light-absorbing agents in modern

photolithographic processes, e.g. in the manufacture of semiconductors, wherein the photoresist layers are exposed to the abovementioned wavelengths but wherein shorter wavelengths, e.g. 193 nm, are particularly desirable for the production of patterns with ever decreasing linewidths. In addition the silyl alkyl esters according to the invention exhibit absorption over an approx. 10nm-wide range around the wavelengths 248 nm and 193 nm, which is also highly significant for photolithographic applications. Absorbing compounds with narrow absorption peaks, e.g. with a width of less than 2 nm around the abovementioned wavelengths, have proved less suitable.

It was also surprising that silyl alkyl esters according to the invention, in which in Formula (I) $n = 3$ and R_4 and R_5 each represent hydrogen, i.e. the silyl group is attached to the ester group via a propylene group, both in the case of compounds based on an anthracene and on a phenanthrene radical, have higher hydrolytic and thermal stability than corresponding compounds with $n = 1$.

The composition according to the invention contains at least one of the silyl alkyl esters according to the invention described above, and at least one further reactive silane. For example, such compositions are suitable as starting materials for the preparation of coatings used in photolithographic processes, e.g. in semiconductor production, which are intended to avoid interference caused by reflection of transmitted light.

The reactive silane is preferably selected from alkoxysilanes and halogen silanes. Suitable alkoxysilanes are for example selected from triethoxysilane (HTEOS), tetraethoxysilane (TEOS), methyltriethoxysilane (MTEOS), dimethyldiethoxysilane, tetramethoxysilane (TMOS), methyltrimethoxysilane (MTMOS), trimethoxysilane, dimethyldimethoxysilane, phenyltriethoxysilane (PTEOS), phenyltrimethoxysilane (PTMOS), diphenyldiethoxysilane and diphenyldimethoxysilane.

Preferred halogen silanes are chlorosilanes, e.g. trichlorosilane, methyltrichlorosilane, ethyltrichlorosilane, phenyltrichlorosilane, tetrachlorosilane, dichlorosilane, methyldichlorosilane, dimethyldichlorosilane, 5 chlorotriethoxysilane, chlorotrimethoxysilane, chloromethyltriethoxysilane, chloroethyltriethoxysilane, chlorophenyltriethoxysilane, chloromethyltrimethoxysilane, chloroethyltrimethoxysilane and chlorophenyltrimethoxysilane. It is yet more preferred that the reactive silane is selected from 10 triethoxysilane (HTEOS), tetraethoxysilane (TEOS), methyltriethoxysilane (MTEOS) tetramethoxysilane (TMOS), methyltrimethoxysilane (MTMOS), phenyltriethoxysilane (PTEOS) and phenyltrimethoxysilane (PTMOS). Tetraethoxysilane (TEOS) and/or methyltriethoxysilane (MTEOS) are most preferred.

15 In a preferred embodiment the composition according to the invention is present in the form of a solution. Consequently the composition preferably contains a solvent or a solvent mixture. It is preferred that the solvent or solvent mixture 20 contains at least one component selected from water, linear or branched alkyl alcohol having 1 to 6 carbon atoms, linear or branched ketone having 1 to 6 carbon atoms, linear or branched carboxylic acid ester having 1 to 6 carbon atoms and linear or branched carboxylic acid amide having 1 to 6 carbon 25 atoms. It is yet more preferred that the solvent or solvent mixture contains at least one component selected from water, acetone, 1-propanol, 2-propanol, butanol, methylisobutylketone, methoxypropanol, propoxypropanol, ethyl acetate and propyl acetate. The most preferred are water, 30 acetone, 1-propanol, 2-propanol and/or butanol. By "butanol" is also meant all isomers of 1-butanol, e.g. 2-butanol or 2-methyl-1-propanol.

35 Furthermore, the composition according to the invention preferably contains an aqueous solution of at least one protonic acid and/or an aqueous solution of at least one acid anhydride. An aqueous solution of at least one protonic acid, e.g. an aqueous nitric acid solution (e.g. 0.1 M-aqueous

nitric acid solution) is preferred. Examples of other suitable protonic acids and acid anhydrides are acetic acid, formic acid, phosphoric acid, hydrochloric acid and acetic acid anhydride.

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Suitable amounts of the above mentioned components in the composition according to the invention, which can be selected independently of one another, are as follows:

10 - 0.5 to 10 wt.-% (more preferably 1 to 4 wt.-% and yet more preferably 2 to 3 wt.-%) silyl alkyl ester according to the invention,

15 - 1 to 30 wt.-% (more preferably 5 to 20 wt.-% and yet more preferably 5 to 15 wt.-%) reactive silane,

- 60 to 98 wt.-% (more preferably 80 to 95 wt.-% and yet more preferably 85 to 90 wt.-%) solvent or solvent mixture and

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- 10^{-5} to $5 \cdot 10^{-3}$ wt.-% (more preferably 10^{-4} to 10^{-3} wt.-% and yet more preferably 10^{-4} to $3 \cdot 10^{-4}$ wt.-%) protonic acid and/or acid anhydride.

25 Usually, the at least one protonic acid and/or the at least one acid anhydride is added in the form of an aqueous solution, e.g. 0.001 to 0.5 wt.-%, preferably 0.01 to 0.1 wt.-% and yet more preferably 0.01 to 0.03 wt.-% of a 1 wt.-% aqueous solution of these components.

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In certain cases it may be an advantage for the composition according to the invention to contain auxiliary agents, e.g. wetting agents. Suitable wetting agents are alkyl and aryl sulfonic acids, alkyl and aryl carboxylic acids, partially fluorinated or perfluoro alkyl and aryl sulfonic acids and partially fluorinated or perfluoro alkyl and aryl carboxylic acids each having a long-chain as well as their respective salts. Preferably, the amount of auxiliary agents in the

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composition according to the invention is 0.001 to 1 wt.-% and more preferably 0.01 to 0.1 wt.-%.

5 The amounts of the components mentioned, given in wt.-% relate to the weight of the total composition.

10 The composition according to the invention is preferably produced in such a way that the starting materials are added in any order, and optionally mixed/homogenized, using methods known to those skilled in the art. For example, only silyl alkyl ester according to the invention, reactive silane and solvent or solvent mixture may be mixed first, and the aqueous solution of at least one protonic acid and/or aqueous solution of at least one acid anhydride added subsequently.

15 In the method according to the invention, for the preparation of a polysiloxane composition, particularly a polysiloxane solution, the composition described above is provided and the silyl alkyl ester is condensed with the reactive silane. 20 Generally, a linear and/or cross-linked polyorganosiloxane is formed, which in the following is also referred to as glass matrix. The condensation of the silyl alkyl ester with the reactive silane is preferably controlled in such a way that polysiloxane with a low molecular weight and degree of cross-linking is formed, which is dissolved or suspended in the solvent or solvent mixture.

30 In a further preferred embodiment, the condensation of the silyl alkyl ester with the reactive silane is carried out by heating. For this purpose the composition according to the invention is preferably heated for 1 to 24 hours, more preferably 1 to 10 hours and yet more preferably 2 to 6 hours at a temperature of 20 to 100°C, more preferably 40 to 80°C and yet more preferably 60 to 80°C. It is also possible simply to heat the composition according to the invention under reflux, so that the required temperature range is determined by the boiling point of the solvent or solvent mixture used. 35 Depending on the proportions of the individual components in

the composition according to the invention, different temperatures and reaction times are required, in order to obtain preferably short-chained and slightly cross-linked polyorganosiloxane. This can be optimized by skilled persons in a few steps. The desired degree of cross-linking can be monitored, e.g. by viscosity measurements.

If the polysiloxane composition according to the invention is applied on a substrate, e.g. a semiconductor wafer, a hard (glass-like) coating can be produced by heating, in which polysiloxane with a higher molecular weight and degree of cross-linking is present.

The light-absorbing silyl alkyl ester according to the invention can either be incorporated in the interstices of the glass matrix or chemically bonded to the polysiloxane. It is assumed that the accessible reactive group of the compounds according to the invention enables chemical bonding to the polysiloxane and this has a particularly advantageous effect on the anti-reflective properties of the coatings.

It may possibly also be advantageous, during preparation of the polysiloxane composition to add a diluting solvent during or after the condensation. This is because, by varying the amount of solvent (including diluting solvent) in the polysiloxane composition, it is possible to adjust the thickness of a coating applied to a substrate. Suitable diluting solvents are methanol, ethanol, 2-propanol, butanol, acetone, propyl acetate, ethyl lactate, propylene glycol propyl ether, diacetone alcohol and methoxypropanol. Diluting solvents with high boiling points are preferred, such as ethyl lactate and propylene glycol propyl ether, presumably because these reduce the probability of bubble formation in the coatings. On the other hand, solvents with lower boiling points lead to cavities if they are enclosed under a cross-linked top layer and consequently again escape due to burning process required for hardening.

Instead of the above single-stage method it is also possible to mix the above mentioned components of the composition according to the invention and optionally diluting solvent in stages in any order, and to condense after any, several, or
5 each individual addition stage.

It may possibly also be necessary to separate solid, insoluble components from the polysiloxane composition, e.g. using standard filtration techniques, before they are
10 available for final use.

The composition according to the invention and the polysiloxane composition according to the invention, which is obtainable by the method described above, can be used
15 particularly for the preparation of a coating on any substrate, for instance a semiconductor device, a silicon wafer, a glass plate, a metal plate and the like. Any coating method known to persons skilled in the art is suitable for the application, e.g. the spin method, roller-coater method,
20 dip and pull-up method, spray method, screen printing method and brushing method. The composition and the polysiloxane composition according to the invention is preferably applied to the substrate and the substrate with said composition or polysiloxane composition applied thereon is heated. In
25 semiconductor production, spin methods are typically used for the application. Such methods include a first spin to apply the coating materials and a second spin to form a layer with a specific thickness. This is followed by at least one thermal burning stage to produce the hard (glass-like) layer.
30 Typical processes operate with the second spin at 1000 to 4000 rpm for approx. 20 seconds and with two or three burning stages at temperatures of 80 to 300°C for approx. 1 minute in each case.

35 The layers thus produced exhibit light-absorbing and anti-reflective properties. In addition they are suitable to planarize the uneven topography of a substrate or a subjacent layer so that variations in the thickness of the layer or

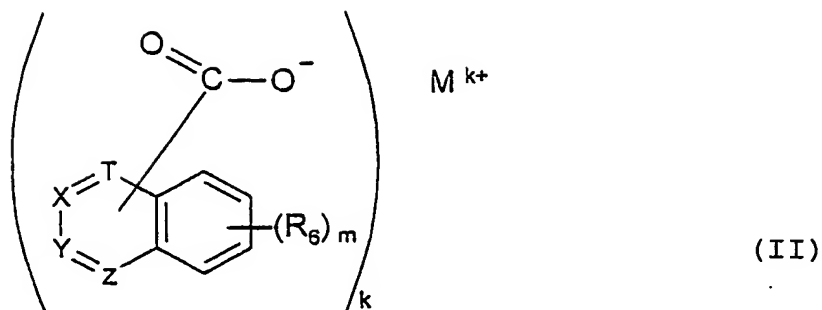
layers lying on top can be reduced.

The method according to the invention for the photolithographic production of an integrated circuit pattern on a substrate comprises the steps of:

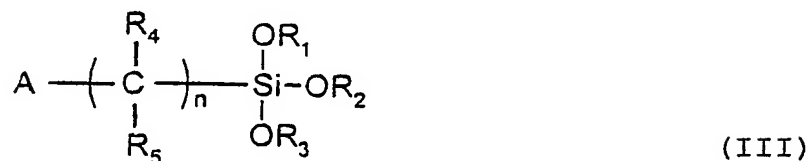
- (a) radiating a stack through a mask, said stack comprising the lower substrate, an upper photoresist layer, at least one anti-reflective layer formed from the composition or from the polysiloxane composition according to the invention, and optionally at least one further layer to be structured, each being located between said substrate and said photoresist layer,
- (b) developing the exposed stack to produce openings in said photoresist layer,
- (c) etching a first pattern through said openings in said at least one anti-reflective layer and optionally at least one further layer to be structured, and
- (d) stripping off said photoresist layer, at least one anti-reflective layer and optionally at least one further layer to be structured to produce the integrated circuit pattern.

Preferably, the substrate is a semi-conductor device or a silicon wafer. It is further preferred that in step (a) the stack is radiated with ultraviolet light having a wavelength of less than 260 nm, particularly 193 nm or 248 nm. Suitable materials and methods to perform the radiating, developing, etching and stripping steps are well-known to those skilled in the art. For further details reference is made to the aforementioned WO-A-00/77575.

The method according to the invention for the preparation of a silyl alkyl ester of Formula (I) comprises the reacting of a carboxylic acid salt of Formula (II)



in which k is an integer from 1 to 4 and M is a metal,
with an organosilicon compound of Formula (III)



5 in which A represents a nucleofugal leaving group.

In the carboxylic acid salt of Formula (II), the metal M is preferably selected from the groups Ia, IIa, IIIa, IVa, Ib, IIb, IVb and VIIb of the periodic table. M is particularly a
10 monovalent metal, i.e. k = 1. It is preferred that M is selected from metals of the Group Ia, more preferably from lithium, sodium and potassium and yet more preferably from sodium and potassium, with potassium being the most preferred.

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R₆ is a substituent selected from halogen, alkyl, aryl,

heteroaryl, hydroxy, alkoxy, aryloether, substituted and unsubstituted amino group, carboxy group, carboxylic acid ester group, carboxylic acid amide group, sulfonic acid group, sulfonic acid ester group, sulfonyl, thio, thioether and nitro. Preferred substituents are halogen, alkyl, aryl, hydroxy, alkoxy and aryloether, with halogen being particularly preferred.

m is an integer from 0 to 4, preferably from 0 to 2 and particularly from 0 to 1, with m = 0 being most preferred.

Furthermore, in Formula (II), T, X, Y and Z each represent carbon. On one of the bonds formed by these carbon atoms, T-X, X-Y, or Y-Z, a further benzo group, which is substituted m-fold with R₆ or is unsubstituted, is condensed to form a trinuclear aromatic ring system. The condensation with the benzo group, which is substituted m-fold with R₆ or unsubstituted, can be either linear on the X-Y bond, or angular on either the T-X or Y-Z bond. Accordingly, the trinuclear condensed aromatic hydrocarbon radical is an anthracene or phenanthrene moiety.

The groups already mentioned above can be used as substituent R₆ of the condensed benzo group on the T-X, X-Y or Y-Z bonds. These can be selected independently of one another for the respective rings of the trinuclear condensed aromatic hydrocarbon radical.

The carboxylate group is substituted on the middle ring, i.e. either on the 9- or on the 10-carbon atom of the anthracene or phenanthrene moiety. The carboxylate group is preferably substituted on the 9-carbon atom.

The other respective position of the middle ring can likewise carry a substituent which can preferably be selected from the radicals already listed for the substituent R₆. It is preferred that the anthracene or phenanthrene moiety is substituted on the 10-carbon atom.

It is yet more preferred that an unsubstituted benzo group is condensed either on the X-Y bond or on either the T-X or Y-Z bond, and in each case $m = 0$. This means that the whole
5 condensed aromatic ring system carries no substituents, with the exception of the carboxylate group on the 9- or 10-carbon atom.

Thus, an anthracene or phenanthrene radical, which is
10 substituted on the respective 9-carbon atom with the carboxylate group, is most preferred as trinuclear condensed aromatic hydrocarbon radical in the carboxylic acid salt of Formula (II).

15 Particularly preferred carboxylic acid salts of Formula (II) are selected from 9-anthracene carboxylic acid-sodium salt, 9-anthracene carboxylic acid-potassium salt, 9-phenanthrene carboxylic acid-sodium salt and 9-phenanthrene carboxylic acid-potassium salt.

20 In the organosilicon compound of Formula (III) A represents a nucleofugal leaving group. Preferably, A is selected from halogen, hydroxy, alkoxy, aryloxy, substituted and unsubstituted amino group, thio, thioether, carboxylic acid
25 ester group (e.g. acetate or trifluoroacetate), alkyl sulfonate (e.g. methane sulfonate or mesylate) and aryl sulfonate (e.g. 4-methylbenzenesulfonate or tosylate). It is preferred that the nucleofugal leaving group A represents halogen, more preferably chlorine and/or bromine, with
30 chlorine being most preferred. In general n is an integer from 1 to 10, preferably 1 to 5, more preferably 1 to 3, and most preferably 3. The groups which can be used as R_1 , R_2 , R_3 , R_4 and R_5 radicals are those already discussed in connection with the silyl alkyl esters according to the invention.

35 According to a preferred embodiment of the method according to the invention, the carboxylic acid salt of Formula (II) is reacted with the organosilicon compound of Formula (III) in a

solvent or solvent mixture, from which the metal salt of the formula MA_k formed is precipitated. In addition, it is particularly advantageous if the desired final product, the silyl alkyl ester according to the invention, remains dissolved/suspended in the solvent or solvent mixture. All the usual solvents can be used for this. However, it is preferred that the solvent or solvent mixture contains at least one component selected from dipolar aprotic solvents, with N,N-dimethylformamide, N,N-dimethylacetamide, N-methylpyrrolidone, dimethylsulfoxide, 1,3-dimethyl-2-imidazolidinone, tetramethyl urea and tetrahydrothiophene-1,1-dioxide (Sulfolan) being more preferred. Most preferred are N,N-dimethylformamide and/or N,N-dimethylacetamide.

The molar ratio of the carboxylic acid salt (II) to the organosilicon compound (III) is preferably 1:1 to 1:2 and more preferably 1:1. Preferably, the reaction of the compounds (II) and (III) is carried out for 0.5 to 15 hours, more preferably 1 to 4 hours at temperatures of 60 to 200°C, more preferably 100 to 140°C.

Surprisingly, using the above method, it was possible to obtain silyl alkyl esters with a greater purity and yield than was possible with conventional processes. As, during reaction of the carboxylic acid salt of Formula (II) with the organosilicon compound of Formula (III) a metallic salt which is insoluble in the solvent or solvent mixture is formed (e.g. an alkali metal halogenide), this can be continuously removed from the reaction vessel during the reaction, and thus the reaction equilibrium can be shifted practically completely to the product side so that the yield is quantitative.

To isolate the desired final product, the reaction mixture can be worked up with usual methods. A very high purity of the final product requires, for example, the removal of mechanical impurities or impurities caused by metals in the ppm-range. This can optionally be carried out by vacuum

distillation, more preferably by distillation techniques for the preparation of extremely pure products known to those skilled in the art, particularly thin film or molecular evaporation (e.g. using rotating strip or spinning band
5 columns), or by recrystallization. Hydrocarbons or hydrocarbon mixtures are suitable for recrystallization, with hexane, heptane, octane, cyclohexane and methylcyclohexane being preferred.

- 10 The preparation of the silyl alkyl esters according to the invention suitably takes place in an inert gas atmosphere (e.g. nitrogen) to reduce oxidation by atmospheric oxygen (e.g. of the anthracene/phenanthrene radical to an anthraqui-
15 none/phenanthrenequinone radical) and humidity, which hydrolyses the ester group. In addition, it is preferred to work in the absence of light, as, for example, anthracene compounds can dimerise during exposure as the result of a photochemical reaction.
- 20 The following examples illustrate the present invention. Unless otherwise indicated, all percentages relate to weight.

Examples

In principle all syntheses and working-up steps were carried out under inert gas (e.g. nitrogen) and exclusion of humidity and light. The reagents used were dried before use. The products obtained were kept under inert gas in the absence of humidity and light.

10 **Example 1a: Synthesis according to the invention of 9-anthracene carboxylic acid (triethoxysilyl) methyl ester**

5 g 9-anthracene carboxylic acid-potassium salt were suspended in 50 ml N,N-dimethylformamide. 5 ml chloromethyl-triethoxysilane were added. The mixture was stirred at 100°C for 1 hour, then cooled to 20°C, drawn off by suction, and the residue washed with 60 ml n-heptane. The combined filtrates were concentrated under vacuum, 50 ml n-heptane were added, stirred for 15 min., then drawn off by suction, the residue washed with n-heptane and the combined filtrates concentrated under vacuum. The residue was dried under vacuum at 0.01 mbar. The yield was quantitative. Purity was above 95% (GC). Light-yellow solid; Fp.: 63°C; Bp: 190°C/0.01 mbar.

25 **Example 1b: Synthesis of 9-anthracene carboxylic acid (triethoxysilyl) methyl ester (comparative)**

9-anthracene carboxylic acid (triethoxysilyl) methyl ester was produced according to Example 14 on page 18 of WO 00/77575 A1. A dark-brown oily liquid was obtained in a yield of approx. 53%. The NMR analysis of this oil gave a yield of approx. 25% (relative to the oil) 9-anthracene carboxylic acid (triethoxysilyl) methyl ester.

35 A comparison of both methods shows that the preparation according to the invention is superior in terms of the yield and purity of the final product as it yields a light-yellow solid with a melting point of 63°C and a content of the

desired compound of over 95%. To achieve such a yield and purity, only a filtration and removal of solvent is necessary. The working up of the reaction mixture is thereby simplified compared with the process known from the state of the art in which, in addition to the decanting and removal of the solvent, a time-consuming column-chromatographic purification must be carried out. To achieve an even higher purity, a further purification can be carried out through recrystallization or distillation in the process according to the invention.

Example 2: 9-phenanthrene carboxylic acid(triethoxysilyl) methyl ester

5 g 9-phenanthrene carboxylic acid-potassium salt were suspended in 50 ml N,N-dimethylformamide. 5 ml chloromethyl-triethoxysilane were added. The mixture was stirred at 120°C for 1 hour, then cooled to 20°C, drawn off by suction, and the residue washed with 60 ml n-heptane. The combined filtrates were concentrated under vacuum. 50 ml n-heptane were added, stirred for 15 min., then drawn off by suction, the residue washed with n-heptane and the combined filtrates concentrated under vacuum. The residue was dried under vacuum at 0.01 bar. The yield was quantitative. Purity was above 95% (GC). Light-yellow solid; Fp.: 26°C.

Example 3: 9-anthracene carboxylic acid-3-(triethoxysilyl) propyl ester

5 g 9-anthracene carboxylic acid-potassium salt were suspended in 50 ml N,N-dimethylacetamide. 5.6 ml (3-chloropropyl) triethoxysilane were added. The mixture was stirred at 100°C for 2 hours, then cooled to 20°C, drawn off by suction, and the residue washed with 60 ml toluene. The combined filtrates were concentrated under vacuum, 50 ml toluene were added, stirred for 15 min., then drawn off by

suction, the residue washed with toluene and the combined filtrates concentrated under vacuum. The residue was dried under vacuum at 0.01 mbar. The yield was quantitative. Purity was above 95% (GC). Yellow liquid; Fp.: < 25°C; Bp: 210°C/0.01 mbar.

Example 4: 9-anthracene carboxylic acid-3-(trimethoxysilyl) propyl ester

5 g 9-anthracene carboxylic acid-sodium salt were suspended in 50 ml N,N-dimethylformamide. 4.7 ml (3-chloropropyl) trimethoxysilane were added. The mixture was stirred at 120°C for 2 hours, then cooled to 20°C, drawn off by suction, and the residue washed with 60 ml n-hexane. The combined filtrates were concentrated under vacuum, 50 ml toluene were added, stirred for 15 min., then drawn off by suction, the residue washed with n-hexane and the combined filtrates concentrated under vacuum. The residue was dried under vacuum at 0.01 mbar. The yield was quantitative. Purity was above 95% (GC). Yellow liquid; Fp.: < 25°C; Bp: 205°C/0.01 mbar.